

UTILITY PATENT APPLICATION TRANSMITTAL
(Large Entity)*(Only for new nonprovisional applications under 37 CFR 1.53(b))*Docket No.
1375A1Total Pages in this Submission
39**TO THE ASSISTANT COMMISSIONER FOR PATENTS**Box Patent Application
Washington, D.C. 20231

I am submitting herewith for filing under 35 U.S.C. 111(a) and 37 C.F.R. 1.53(b) is a new utility patent application for an invention entitled:

SHIPPABLE HEAT-TREATABLE SPUTTER COATED ARTICLE AND ZINC CATHODE SPUTTERING TARGET CONTAINING LOW AMOUNTS OF TIN

and invented by:

James J. Finley, Paul A. Medwick and Dennis J. O'ShaughnessyIf a **CONTINUATION APPLICATION**, check appropriate box and supply the requisite information:☒ **Continuation** ☐ **Divisional** ☐ **Continuation-in-part (CIP)** of prior application No.: _____

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Enclosed are:

Application Elements

1. ☒ Filing fee as calculated and transmitted as described below
2. ☒ Specification having 34 pages and including the following:
- a. ☒ Descriptive Title of the Invention
 - b. ☒ Cross References to Related Applications *(if applicable)*
 - c. ☐ Statement Regarding Federally-sponsored Research/Development *(if applicable)*
 - d. ☐ Reference to Microfiche Appendix *(if applicable)*
 - e. ☒ Background of the Invention
 - f. ☒ Brief Summary of the Invention
 - g. ☐ Brief Description of the Drawings *(if drawings filed)*
 - h. ☒ Detailed Description
 - i. ☒ Claim(s) as Classified Below
 - j. ☒ Abstract of the Disclosure

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Application Elements (Continued)

3. ☐ Drawing(s) (when necessary as prescribed by 35 USC 113)
- a. ☐ Formal Number of Sheets _____
- b. ☐ Informal Number of Sheets _____
4. ☒ Oath or Declaration
- a. ☒ Newly executed (original or copy) ☐ Unexecuted
- b. ☐ Copy from a prior application (37 CFR 1.63(d)) (for continuation/divisional application only)
- c. ☒ With Power of Attorney ☐ Without Power of Attorney
- d. ☐ DELETION OF INVENTOR(S)
Signed statement attached deleting inventor(s) named in the prior application,
see 37 C.F.R. 1.63(d)(2) and 1.33(b).
- ☐ Incorporation By Reference (usable if Box 4b is checked)
The entire disclosure of the prior application, from which a copy of the oath or declaration is supplied under
Box 4b, is considered as being part of the disclosure of the accompanying application and is hereby
incorporated by reference therein.
- ☐ Computer Program in Microfiche (Appendix)
- ☐ Nucleotide and/or Amino Acid Sequence Submission (if applicable, all must be included)
- a. ☐ Paper Copy
- b. ☐ Computer Readable Copy (identical to computer copy)
- c. ☐ Statement Verifying Identical Paper and Computer Readable Copy

Accompanying Application Parts

8. ☒ Assignment Papers (cover sheet & document(s))
9. ☐ 37 CFR 3.73(B) Statement (when there is an assignee)
10. ☐ English Translation Document (if applicable)
11. ☐ Information Disclosure Statement/PTO-1449 ☐ Copies of IDS Citations
12. ☐ Preliminary Amendment
13. ☒ Acknowledgment postcard
14. ☐ Certificate of Mailing
- ☐ First Class ☐ Express Mail (Specify Label No.): _____

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Accompanying Application Parts (Continued)

15. ☐ Certified Copy of Priority Document(s) (if foreign priority is claimed)

16. ☒ Additional Enclosures (please identify below):

This application claims priority to U.S. Provisional Application Serial Nos. 60/084,720 and 60/085,129 filed May 8, 1998, and May 12, 1998, respectively.

Fee Calculation and Transmittal

CLAIMS AS FILED

For	#Filed	#Allowed	#Extra	Rate	Fee
Total Claims	37	- 20 =	17	x \$18.00	\$306.00
Indep. Claims	8	- 3 =	5	x \$78.00	\$390.00
Multiple Dependent Claims (check if applicable) <input type="checkbox"/>					\$0.00
BASIC FEE					\$790.00
OTHER FEE (specify purpose)					\$0.00
TOTAL FILING FEE					\$1,486.00

- ☐ A check in the amount of _____ to cover the filing fee is enclosed.
- ☒ The Commissioner is hereby authorized to charge and credit Deposit Account No. 16-2025 as described below. A duplicate copy of this sheet is enclosed.
- ☒ Charge the amount of \$1,486.00 as filing fee.
- ☒ Credit any overpayment.
- ☒ Charge any additional filing fees required under 37 C.F.R. 1.16 and 1.17.
- ☐ Charge the issue fee set in 37 C.F.R. 1.18 at the mailing of the Notice of Allowance, pursuant to 37 C.F.R. 1.311(b).


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Dated: April 29, 1999

Pittsburgh, Pennsylvania
CC:

**SHIPPABLE HEAT-TREATABLE SPUTTER COATED ARTICLE AND
ZINC CATHODE SPUTTERING TARGET CONTAINING LOW AMOUNTS OF TIN**

CROSS-REFERENCE TO RELATED APPLICATIONS

5 This application claims priority to U.S. Provisional
Application Serial Nos. 60/084,720, and 60/085,129 filed
May 8, 1998, and May 12, 1998, respectively.

FIELD OF THE INVENTION

10 This invention relates, in general, to a zinc
cathode sputtering target having low weight percent of tin, to
a shippable, heat treatable, low emissivity coating, which
coating may be made using the target, to coated articles and
to methods of making the coated articles.

15 **DISCUSSION OF THE PRESENTLY AVAILABLE TECHNOLOGY**

U.S. Patent No. 4,610,771 ("U.S.P.N. '771")
discloses an antireflective sputtered metal oxide film
deposited using a zinc/tin alloy target. U.S.P.N. '771 in
20 column 3, line 26, to column 4, line 12, discusses the use of
the alloy target to deposit a zinc stannate film having, in
general, oxides of zinc and tin preferably in proportions of
10 to 90 percent zinc and 90 to 10 percent tin.

Although the zinc/tin alloy target disclosed in
25 U.S.P.N. '771 to deposit the zinc stannate film is acceptable,
there are limitations. More particularly, in a low E coating
such as the type described in U.S.P.N. '771, an infrared
reflecting metal film or layer, e.g., silver is deposited on
the zinc stannate film. A silver film deposited on a
30 sputtered zinc stannate film has a higher electrical
resistivity and higher emissivity than a silver film deposited
on a sputtered zinc oxide film or layer. More particularly,
U.S. Patent No. 5,821,001 ("U.S.P.N. '001") discloses a silver
film deposited on a zinc oxide film; the atoms of the silver
35 film deposit in a form characterized by a low electrical
resistivity which provides the silver film with a low
emissivity. In depositing the zinc oxide film the process

parameters are selected to deposit a zinc oxide layer with a suitable crystallinity or preferential crystal growth for favorably affecting deposition of the silver atoms of the silver film.

5 Sputtering zinc in a reactive atmosphere, e.g., oxygen, to provide a zinc oxide film over which a silver film having a low electrical resistivity is deposited has drawbacks. For example, it is difficult to reactively sputter a pure zinc target, i.e., a target of about 100% zinc metal,
10 in a reactive atmosphere such as oxygen for reasons discussed in more detail below.

U.S.P.N. '001 also discloses a heat treatable low emissivity film. The thickness of the primer layers, e.g., titanium films, may be increased to provide enhanced
15 mechanical durability, i.e., improve shear resistance. The shear resistance test consists of applying 20 successive strokes of a cloth wetted with deionized water against the coated surface of glass, followed by visual examination of the tested area. Depending on the appearance of the tested area,
20 letter grades of D-, D, D+... A, A+ are assigned to the coating; then, for numerical analysis, assignments of 5 to D-,
10 to D, 55 to A, and 60 to A+ are made. If a coating shows no signs of shear, not even barely visible scratches, then it receives a maximum rating of 60. Coatings that display
25 uniform shear and delamination at any interface of the multi-layer coating within the test area receive a failing rating of zero. Other levels of performance receive intermediate scores. This method of coating durability characterization has been found to correlate well with field performance of the
30 coating. The drawback with using thick primer layers is that the coating stack after heating, e.g., glass tempering or glass bending operations, is likely to have a hazy appearance when viewed using a dark room, flood-light haze test. In the dark room, floodlight haze test, the coated specimen is viewed
35 in reflection in a dark room at various viewing angles relative to a spotlight, in order to find the geometry yielding maximum scattering of light, or, in other words,

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haze, possible from the coating. If there is no geometry that can make haze observable, an A+ rating is assigned to the sample. Very poor samples receive D-. For purposes of numerical analysis, the letter grades are given values of 5 to 60, as described above for the shear test. Lower haze corresponds to higher numerical values.

As can be appreciated by those skilled in the art of making sputtered coatings, it would be advantageous to provide a zinc target that may be sputtered in a reactive atmosphere without the drawbacks of the presently available zinc targets and to provide low emissivity coated articles that have mechanical durability so that the coated article may be shipped and subsequently heated without the heated coating having haze.

The disclosures of U.S.P.N.'s '771 and '001 are hereby incorporated by reference.

SUMMARY OF THE INVENTION

This invention relates to a sputter cathode target having tin preferably in an amount greater than zero and less than 10 weight percent of the total weight of the target material, and zinc in an amount preferably less than 100 weight percent and more than 90 weight percent of the total weight of the target material. Hereafter, unless indicated otherwise, the term "weight percent" means the weight percent of the total weight of the target material.

As used herein, "a zinc stannate film," e.g., of the type discussed in U.S.P.N. '771 is an oxide of an alloy of zinc and tin. The cathode used is made of an alloy of zinc and tin. A "zinc oxide, tin oxide film," is a film having oxides of tin and zinc. The cathode used to sputter the zinc oxide, tin oxide film is made of zinc having additions of tin as is discussed in detail below.

In one embodiment of the invention a coating stack has a zinc stannate film deposited on a glass substrate, a zinc oxide, tin oxide film deposited on the zinc stannate film; an infrared reflecting film, e.g., silver, deposited on

the zinc oxide, tin oxide film; a primer layer, e.g., a titanium metal film, deposited on the infrared reflecting film; a zinc oxide, tin oxide film deposited on the primer film, a zinc stannate film deposited on the zinc oxide, tin oxide film; an infrared reflecting film deposited on the zinc stannate film; a primer layer deposited on the infrared reflecting layer; a zinc oxide, tin oxide film deposited on the primer layer, a zinc stannate film deposited on the zinc oxide, tin oxide film; and a protective layer, e.g., a titanium metal film or a titanium oxide film, deposited over the zinc stannate film. In another embodiment of the invention, a zinc oxide film is used in place of the zinc oxide, tin oxide film. In a further embodiment of the invention, a zinc stannate film is used in place of the zinc oxide, tin oxide film. When a zinc stannate film is used in place of the zinc oxide, tin oxide film, the zinc stannate films differ in composition by at least 5 weight percent. For example, when one of the zinc stannate films is about 50 weight percent zinc and 50 weight percent tin, the other zinc stannate film is about 10-45 or 55-90 weight percent zinc and 55-90 or 45-10 weight percent tin. In a still further embodiment of the invention, a first deposited zinc stannate film is 50 ± 10 weight percent zinc and 50 ± 10 weight percent tin. The second deposited or overlying zinc stannate film has tin equal to or greater than 10 weight percent and less than 40 weight percent and preferably 20 weight percent, and has zinc equal to or less than 90 weight percent and greater than 60 weight percent and preferably 80 weight percent. Overlying zinc stannate films having 90 weight percent zinc and 10 weight percent tin have been used.

The above-described coating stack has mechanical and chemical durability. The coatings of the instant invention in addition to having mechanical and chemical durability which make them acceptable for shipment can be heat treated with the heat treated coating having reduced haze. The reduction in haze after heating is accomplished by selecting metal primer thicknesses which is discussed below. The procedure for

determining haze was discussed above. As used herein a reduction in haze is a numerical increase of about 10. An advantage of the coatings of the instant invention is that a coating of the invention may be deposited on a substrate, the coated substrate shipped to a fabricating facility where the coated substrate is heat-treated, e.g., heated to temperatures up to about 1350°F (732°C). The coating of the invention has particular application in the making of solar control automotive windshields. Sheets of glass having a low emissivity coating of the invention are coated in one facility and then shipped to another facility where the coated glass sheet is processed, e.g., into an automobile windshield.

DISCUSSION OF THE INVENTION

For a full appreciation of the various embodiments of the invention, the embodiments will be discussed separately and then combined. The first embodiment of the invention to be discussed is the use of a zinc cathode having low amounts of tin to improve the deposition of zinc oxide films and the benefits of a zinc oxide film having low amounts of tin.

The embodiments of the invention relating to zinc sputtering target having low amounts of tin and to a method of reactively sputtering such zinc targets to sputter deposit a zinc oxide, tin oxide film will be discussed. As will be appreciated, the invention contemplates using the target of the instant invention to deposit a zinc oxide, tin oxide film over or under an infrared reflective film, e.g., gold, silver or aluminum film, other metal films, e.g., primer films such as a titanium metal, or ceramic films, or other dielectric films.

The sputtering cathode target in one embodiment of the invention has greater than zero weight percent and less than 10 weight percent tin, and less than 100 weight percent and more than 90 weight percent zinc to improve the emissivity of an infrared reflective film, e.g., a silver film, deposited on the zinc oxide, tin oxide film and to improve the sputtering of the zinc cathode target having low amounts of

tin. The cathode target of the instant invention may also be defined as having greater than 0 and less than 10 weight percent tin with the majority of the balance zinc, or as having less than 100 weight percent and more than 90 weight percent zinc with the majority of the balance tin. Before discussing this embodiment of the invention, the drawbacks and/or limitations of reactively sputtering a zinc cathode target without tin and limitations of zinc oxide films deposited using such cathodes are discussed for a better appreciation of the invention.

Typically, a metal target is reactively sputtered in a gas such as oxygen, or gas mixture of oxygen with other gases such as nitrogen, argon or helium. Sputtering in a gas mixture will result in a higher sputtering rate for a stable process but requires more control to maintain the process stability, e.g., requires controlling flow rate of two gases. Because of the higher sputtering rate, it is preferred to sputter in a gas mixture rather than in oxygen alone. In either case the resulting coating is a metal oxide, e.g., zinc oxide when a zinc target is used.

Zinc oxide is a common dielectric material that is used as a high refractive index film in coating stacks having low emissivity. In the flat glass industry these coatings are usually applied by horizontal vacuum coaters using high power supplied from a direct current source to energize cathodes that sputter layers onto glass substrates. Higher coater throughput requires higher power density to the cathode target. This in turn increases the tendency for the targets to arc, particularly during reactive sputtering.

During reactive sputtering the frequency of arcing of the zinc cathode target increases with time, and debris in the form of powder and flakes builds up on the target surface and adjacent areas. Debris in the form of powders, flakes and splatter eventually falls onto the surface of the substrate being coated resulting in unacceptable coated product. Further the arcing increases to a point where the process becomes unstable. In addition, areas of the target surface

have a tendency to blacken over time. These blackened areas are nonconductive, thus limiting the sputtering rate and leading to non-uniformity in the coating.

The arcing and debris buildup to some degree may be reduced by periodic sputter cleaning as is known by those skilled in the art of sputter coating. One technique for sputter cleaning is to periodically sputter the targets for a certain time period in an inert gas, such as argon or helium, which sputters the target as a metal. Sputter cleaning to some degree removes the oxide buildup on the target surface that causes arcing. The black areas on the target may not be reduced by sputter cleaning. The debris and arcing, however, continually degrade the target, and after a period of time, the coater downtime increases and, consequently, there is less production time for the coater. Zinc targets because of their tendency to frequently arc during sputtering are difficult to sputter clean and require longer and more frequent cleaning.

In the practice of the invention, tin is added to a zinc target to reduce if not eliminate the above drawbacks, e.g., reduce the amount of flaking debris, reduce powdery buildup on the target, minimize arcing, and minimize, if not eliminate, blackening of the target surface areas. The amount of debris and degrading of the zinc target of the instant invention is considerably less with time than for a pure zinc target. As a result, less periodic target cleaning is required with less duration of the cleaning in an inert gas.

As previously discussed, U.S.P.N. '771 discloses a zinc-tin alloy target to deposit a zinc stannate film, i.e., an oxide of a zinc tin alloy having 10 to 90 weight percent zinc and 90-10 weight percent tin. The zinc-tin alloy cathode target of U.S.P.N. '771 provides a zinc stannate film that has better chemical durability than a zinc oxide film. Further, the zinc-tin alloy cathode target has less arcing and minimal debris buildup, e.g., there is no measurable powder buildup. As is well-known in the art of sputter coating, zinc oxide films readily dissolve in acid and base solutions; zinc stannate films have reduced solubility in acid or base.

Although the properties of the zinc oxide, tin oxide film of the instant invention have not been studied in great detail, it is believed that the following will occur. As the tin approaches zero weight percent, the chemical durability of the deposited film decreases, and the problems associated with sputtering a zinc target in a reactive atmosphere increase. As the weight percent of tin approaches ten, the chemical durability of the zinc oxide film increases, and the problems associated with sputtering a zinc target in a reactive atmosphere decrease. The electrical resistivity of a silver film deposited on the zinc oxide, tin oxide film is expected to be similar to a silver film deposited on a zinc oxide film. In the practice of this embodiment of the invention, a zinc target having tin greater than zero and less than ten weight percent is a useable range; 0.5 to 9.5 weight percent of tin is a practical range; 4 to 8.5 weight percent of tin is a preferred range, and 5 to 9.5 weight percent of tin is a more preferred range. The weight percent of tin and zinc in a zinc oxide, tin oxide film is expected to be similar to the weight percent of zinc and tin in the target.

As can now be appreciated, as the weight percent of tin in the zinc cathode increases, the chemical durability of the deposited film is expected to increase. Further, the crystalline structure of the zinc oxide film having tin in weight percents greater than zero and less than ten is similar if not identical to the crystalline structure of zinc oxide film having zero weight percent of tin. Further, a zinc stannate film having 60-90 weight percent zinc and 10-40 weight percent tin have similar crystalline structures as zinc oxide. Therefore, it is expected that emissivity of silver film deposited on a zinc oxide film will be similar to a silver film deposited on a zinc stannate film having 60-90 weight percent zinc and 10-40 weight percent tin. At values of less than 60 weight percent zinc, the crystalline structure starts to change and the emissivity and resistivity start to increase. Transmission Electron Microscopy has shown a weak zinc oxide electron diffraction pattern for a zinc stannate

film having 66 weight percent zinc and 34 weight percent tin, and an amorphous structure for zinc stannate film having 47 weight percent zinc and 53 weight percent tin.

The coated articles disclosed in U.S.P.N. '001 and
5 U.S. Patent Application Serial No. 09/023,746 filed
February 13, 1998, in the names of Mehran Arbab, Russell C.
Criss, Gary J. Marietti and Paul A. Medwick for "Coated
Articles" (hereinafter "U.S.P.A. No. 09/023,746") may be made
practicing this embodiment of the instant invention which
10 includes but is not limited to sputtering a cathode target
having 92 weight percent zinc and 8 weight percent tin in an
atmosphere having more than 75% oxygen and the remainder
argon. Further, problems discussed above associated with
sputtering a zinc target are expected to be minimized if not
15 eliminated.

The discussion will now be directed to improving
chemical durability of the coating stack, reduction of haze of
the coating stack and decreased emissivity of the infrared
reflective metal, e.g., silver. "Chemical durability" means
20 the coating is not readily attacked by acid or base solutions.
Reference may be had to U.S.P.N.'s. '001 and '771 for a
discussion on chemical durability. The test for haze was
discussed above.

From the above discussion, the emissivity of the
25 silver film may be reduced by depositing the silver layer on a
zinc oxide film; a zinc oxide, tin oxide film or a zinc
stannate film having zinc in the range of 60 to 90 weight
percent and tin in the range of 10 to 40 weight percent.
Further, from the above discussion, chemical durability of a
30 layer having a zinc oxide film under a silver film and over a
zinc stannate film may be enhanced by adding tin to a zinc
target to provide a zinc oxide, tin oxide film, or a zinc
stannate film. As used herein, "a chemical and electrical
enhancement film" is a zinc oxide, tin oxide film and/or a
35 zinc stannate film having zinc in the range of 60 to 90 weight
percent and tin in the range of 10 to 40 weight percent. The
chemical and electrical enhancement film of the invention may

be used in place of zinc oxide films to enhance chemical durability of the coating stack while obtaining a silver film having low emissivity. By way of illustration, coated articles having a glass substrate/zinc stannate film/zinc oxide film/silver film/titanium metal primer film/zinc oxide film/zinc stannate film/zinc oxide film/zinc stannate film/titanium oxide protective overcoat may be chemically enhanced by using the chemical and electrical enhancement film of the invention for one or more or all of the zinc oxide film(s) of the above coating stack.

Another coated article includes glass substrate/zinc stannate film/zinc oxide film/silver film/titanium metal primer film/zinc oxide film/zinc stannate film/titanium oxide protective overcoat. As can now be appreciated, the chemical and electrical enhancement film of the invention may be substituted for one or more or all of the zinc oxide film(s) of the above coating.

In the practice of the instant invention, a dielectric layer may include a zinc stannate film, and a chemical and electrical enhancement film. Where the chemical and electrical enhancement film is zinc stannate, the difference between the composition of the zinc stannate film and the zinc stannate film of the chemical and electrical enhancement film is at least 5 weight percent. For example, and not limiting to the invention a zinc stannate film having 58 weight percent zinc and 42 weight percent tin may be used with a zinc stannate film (chemical and electrical enhancement film) having 63 to 90 weight percent zinc and 10 to 37 weight percent tin.

The discussion will now be directed to the embodiment of the invention to provide a coating stack that is chemically and mechanically durable and a coating stack that has reduced haze after subjecting the coating stack to elevated temperatures, e.g., but not limiting to the invention, above room temperature and below about 1350°F (732°C). As will be appreciated by those skilled in the art, the invention is not limited to the coatings discussed below

which are presented for illustration purposes only. The following Table 1 provides a few embodiments of coating stacks that may be used in the practice of the invention; however, as will be appreciated, the invention is not limited thereto.

5

TABLE 1

Coating Samples	Substrate	FILM											
		1	2	3	4	5	6	7	8	9	10	11	12
1	X	X	X	X	X		X						X
2	X	X	X	X	X		X		X	X		X	X
3	X	X		X	X	X	X		X	X		X	X
4	X	X		X	X		X		X	X	X	X	X
5	X	X	X	X	X	X	X		X	X		X	X
6	X	X	X	X	X		X		X	X	X	X	X
7	X	X		X	X	X	X		X	X	X	X	X
8	X	X	X	X	X	X	X		X	X	X	X	X
9	X	X	X	X	X	X	X	X	X	X	X	X	X
		A			B			C			D		
		E			F			G			H		
		Layers											

The second column from the left is entitled "Substrate." The material of the substrate is not limiting to the invention and may be made of any material, e.g., glass, fiberglass, plastic, metal, wood or ceramic. The type of articles that are made in the preferred practice of the invention are transparencies for residential and commercial buildings, and land, air, space, above, on and below water vehicles, therefore, the substrate is preferably transparent and made of glass, and flexible and rigid plastics. Glass when used may be clear or tinted and the type of glass is not limiting to the invention. It is expected that the coated article will be subjected to elevated temperatures; therefore, the substrate selected should be able to withstand the elevated temperatures. In our discussion, but not limiting to the invention, the substrates are glass sheets or pieces.

The columns numbered 1-12 are films, and the columns labeled A-H are layers, of coating stacks that incorporate features of the invention. The layers (see the bottom of the Table 1) include at least one film and as shown in the Table 1 up to 3 films. The layers A, D and G are dielectric layers.

The index of refraction of the dielectric films of the Layers A, D and G is preferably greater than the index of refraction of the transparent substrate to anti-reflect the infrared reflection layer. The invention is not limited to the type of

5 dielectric films that may be used in combination with the chemical and electrical enhancement film of the invention.

Dielectric films that may be used in the practice of the invention include but are not limited to zinc oxide, tin oxide, ~~silicon~~ ^{silicon} oxide, silicon oxide, silicon nitride, and ~~silver~~ ^{tin} oxynitride. *7/10 4/29/99 OAM 4/29/99 JJP 4/29/99*

10 It is preferred that films 1, 6 and 11 of the layers A, D and G, respectively are each a zinc stannate film having 52 weight percent zinc and 48 weight percent tin. The films 2, 5 and 7, and 10 of the layers A, D and G respectively may be a zinc

oxide film, or a chemical and electrical enhancement film of 15 the invention. In the following discussion, the substrate is soda-lime-silicate clear glass and has an index of refraction about 1.5. As is known in the art, varying the thickness of the film and layers changes the color of the coated article, or may provide a coating with a neutral color. It is expected

20 that in the practice of the invention, the dielectric layers and/or films have a thickness in the range of 600 ± 500 Angstroms. The thickness of the zinc oxide film, ^{or the} ~~chemical and~~ *7/10 4/29/99 OAM 4/29/99 JJP 4/29/99* electrical enhancement film should be sufficient to affect the crystal structure of the silver film to be deposited thereon.

25 The films 3 and 8 of layers B and E, respectively, are infrared reflecting films and may be of any material that reflects infrared energy, e.g., but not limiting to gold, silver and aluminum. In the practice of the invention silver is preferred. The thickness of the silver is not limiting to 30 the invention and is selected to provide a transparent coating having low emissivity. Silver films having a thickness of $200 \pm 150 \text{ \AA}$ and preferably $100 \pm 25 \text{ \AA}$ may be used in the practice of the invention.

The films 4 and 9 of the layers C and F, 35 respectively, are primer films which have the function of (1) protecting the infrared metal layer from oxidizing during sputtering of the dielectric films, (2) protecting the

infrared reflecting layer during high temperature processing,
(3) reducing haze formation in the coating stack during
heating, and/or (4) providing the coating stack with
mechanical durability for shipping the coated article. Primer
5 films may be any of the type known in the art, e.g., metals
such as titanium or ceramic, of the type disclosed in U.S.
Patent Application Serial No. 09/215,560, filed December 8,
1998, the disclosure of which is hereby incorporated by
reference. In the practice of the invention, the primer layer
10 is preferably titanium.

Film 12 of the layer H is a protective film to
provide additional chemical and mechanical durability for the
coating stack during shipping and storage. The invention is
not limited to the type of protective film and any of the
15 types known in the art may be used, e.g., titanium, titanium
dioxide, silicon oxide, silicon dioxide, silicon aluminum
nitride. Further, more than one protective film may be used.
For example, but not limiting to the invention, a zinc oxide
film over a titanium dioxide film may be used. The thickness
20 of the Layer H is not limiting to the invention; however, the
thickness should be thick enough to provide protection.

Before discussing in detail the Samples of the Table
1, the following background information is provided for a
better appreciate of the invention.

25 Using zinc oxide film, as discussed above, provides
a silver layer having resistivity and emissivity lower than a
silver layer deposited on a zinc stannate layer having less
than about 60 weight percent zinc and more than about 40
weight percent tin.

30 U.S.P.N. '001 discusses increasing the thickness of
the primer layer to enhance the mechanical durability of the
coated article to make the coated article shippable. More
particularly, U.S.P.N. 001 discloses that it has been found
that where the coated article will be exposed to heat
35 treatment during its production, there is a point at which the
primer layer may be made either too thin or too thick. Too
thin a primer layer results in a lack of protection for the

reflective, metallic film from oxidation at high temperature thus rendering the coated article unacceptable for heat treatment and in poor shear resistance which makes the article unsuitable for long distance shipment for later thermal processing. Too thick a primer layer results in the formation of an undesirable haze in the coated article after heat treatment, also rendering it unacceptable for heat treatment. However, a limitation is that these films after heating have haze.

10 It has been determined that by selecting dielectric films and primer layers that a coating stack can be made that has reduced haze after heating. For coated articles that are shipped but not heated, the primer layer should be thick enough to protect the silver during deposition of the
15 overlying dielectric film or layer on the silver layer. *D/O 4/29/99 O.K.M. 4/29/99*
D/O 4/29/99 O.K.M. 4/29/99
~~layers C and F.~~ A thickness of primer layers in the range of about 8 to 12 Angstroms is sufficient. The thickness of the primer layer is increased when the primer layer is to protect the silver during heating of the coated article. A thickness
20 of about $20 \pm 5 \text{ \AA}$ is acceptable.

For a coated article that is shippable and heatable with reduced haze, the thickness of the primer film is adjusted to compliment the dielectric layer or film arrangement. In accordance with the teachings of the
25 invention, primer layer thickness in the range of 18-32 Angstroms (\AA) and preferably 18-40 \AA is acceptable to provide a coating stack with reduced haze after heating. The following Examples illustrate the invention.

In the following discussion, the thickness of the
30 metal primer layer is as deposited. As can be appreciated, the thickness increases after heating, changing a portion of the titanium metal primer film to titanium oxide. A method that will be referred to as the "XRF Method" is discussed in U.S.P.N. '001. In general the XRF Method is used to measure
35 the thickness of metal layers. The XRF Method uses calibrated x-ray fluorescence instrument to measure the weight of the metal per unit area of the coating (namely, in $\mu\text{g}/\text{cm}^2$).

The XRF Method makes the assumption that the metal film is as dense as its bulk form. With this assumption, the metal film's measured weight per unit area is then converted to a thickness in Angstroms, using its bulk density.

5 For completeness sake, it should be noted that sputtered metal films are often less dense than their corresponding bulk metals, so that above described assumption is not always precisely correct, and the XRF Method may in some cases underestimate the thickness of the metal film due to this
10 variation in density. Thus, for the thin metal films, the initial measurement of weight per unit area ($\mu\text{g}/\text{cm}^2$) is more accurate than the corresponding conversion to thickness based upon bulk density. Nonetheless, the XRF Method provides a useful approximation for comparing the relative thicknesses of
15 the layers in coating. ~~Thickness tolerances given herein represent twice the standard deviation of the measurements.~~ *DP 4/29/99*
QAM 4/29/99

In the following discussion, the ~~thickness of the~~ *thickness of the* dielectric layers and/or films are given in ranges. As can be appreciated by those skilled in the art, the ranges are not limiting to the
20 invention and the thickness may be selected to provide a coating stack of a desired color. *DP 4/29/99*
QAM 4/29/99

EXAMPLE 1

This Example 1 is Sample 1 of the Table 1. Sample 1
25 is a coated article that is shippable and heatable. The coating is a high transmittance, low emissivity coated article having a single infrared, reflective layer. Product having the coating stack of Sample 1 is made and the coating stack includes:

30 a clear glass substrate; a dielectric, antireflective layer deposited on the substrate, the layer includes (1) a zinc stannate film having 52 weight percent zinc and 48 weight percent tin (hereinafter referred to as 52-48 zinc stannate film) and having a thickness of $260 \pm 40\text{\AA}$, and
35 (2) a zinc stannate film having 90 weight percent zinc and 10 weight percent tin (hereinafter referred to as 90-10 zinc stannate film) and having a thickness of about $80 \pm 4\text{\AA}$; *DP 4/29/99*
QAM 4/29/99
DP 4/29/99

a silver film having a thickness of about $115 \pm 15\text{\AA}$ deposited on the 90-10 zinc stannate film,

a titanium primer film having a thickness of $24-28\text{\AA}$ deposited on the metallic reflective film;

5 a dielectric, antireflective upper layer deposited on the titanium primer film, the dielectric, antireflective upper film includes a 52-48 zinc stannate film having a thickness of about $230 \pm 40\text{\AA}$ deposited on the primer layer metal oxide, and

10 a titanium oxide layer having a thickness of $36 \pm 8\text{\AA}$ deposited on the 52-48 zinc stannate layer or film.

EXAMPLE 2

This Example 2 is Sample 2 of the Table 1. The coated article has been made and is shippable and heatable with reduced haze. The coated article includes a glass substrate/a layer of a 52-48 zinc stannate film having a thickness of about $230 \pm 40\text{\AA}$ and a zinc oxide film having a thickness of about $80 \pm 40\text{\AA}$; a silver film having a thickness of $110 \pm 10\text{\AA}$; a titanium metal primer film having a thickness of about $18 - 23\text{\AA}$ and preferably 19.5\AA ; a 52-48 zinc stannate film having a thickness of about $20 \pm 40\text{\AA}$; a silver film having a thickness of about $110 \pm 10\text{\AA}$; a metal primer having a thickness of about $18 - 31\text{\AA}$ and preferably 25\AA ; a 52-48 zinc stannate film having a thickness of about $200 \pm 20\text{\AA}$, and a titanium film having a thickness of about $29 \pm 3\text{\AA}$.

EXAMPLE 3

This Example 3 is Sample 3 of the Table 1. The coated article was not made; however, the following coated article is expected to be shippable and heatable with reduced haze. Example 3 includes a clear glass substrate; a dielectric, antireflective base layer deposited on the substrate includes a 52-48 zinc stannate film having a thickness of about $300 \pm 20\text{\AA}$ deposited on the glass substrate; a first silver film having a thickness of about $110 \pm 10\text{\AA}$

deposited on the 52-48 zinc stannate film; a first titanium primer film having a thickness of 18-29Å deposited on the first silver film; a dielectric, antireflective intermediate layer deposited on the first primer film, the intermediate layer includes a zinc oxide film having a thickness of $80 \pm 40\text{\AA}$ deposited on the first primer film, a 52-48 zinc stannate film having a thickness of $740 \pm 40\text{\AA}$ deposited on the ~~52-48 zinc~~ stannate film; a second silver film having a thickness of about $110 \pm 10\text{\AA}$ deposited on the 52-48 zinc stannate film of the intermediate layer; a second titanium primer film having a thickness of about 18-31Å deposited on the second silver film; a dielectric, antireflective upper layer deposited on the second primer film, the dielectric upper layer is a 52-48 zinc stannate film having a thickness of about $200 \pm 20\text{\AA}$; and a titanium metal protective film having a thickness of about 29 \pm 3Å, deposited on the 52-48 zinc stannate film of the dielectric upper layer.

EXAMPLE 4

This Example 4 is Sample 4 of the Table 1. The coated article of Example 4 was made and is shippable and heatable with the heated, coated article having reduced haze. The coated article of this Example 4 includes a clear glass substrate; a 52-48 zinc stannate film having a thickness of $310 \pm 20\text{\AA}$ deposited on the glass substrate; a first silver film having a thickness of $110 \pm 10\text{\AA}$ deposited on the 52-48 zinc stannate film; a first titanium primer having a thickness of 18-29Å and preferably 22.5Å deposited on the first silver film; a 52-48 zinc stannate film having a thickness of about $820 \pm 40\text{\AA}$ deposited on the first titanium film; a second silver film having a thickness of about $110 \pm 10\text{\AA}$ deposited on the 52-48 zinc stannate film; a second titanium film having a thickness of 18-32Å and preferably 21.5Å is deposited on the second silver layer; a zinc oxide film having a thickness of $80 \pm 40\text{\AA}$ deposited on the

second titanium primer layer; a 52-48 zinc stannate film having a thickness of $120 \pm 40\text{\AA}$ deposited on the zinc oxide film and a titanium overcoat film having a thickness of $29 \pm 3\text{\AA}$ over the 52-48 zinc stannate film.

5

EXAMPLE 5

This Example 5 is Sample 5 of the Table 1. The coated article was not made; however, it is expected that the coated article is suitable for shipment and heat treatment with the heated, coated article having reduced haze. The coated article of this Example 5 includes the films and layers similar to Example 3 except Sample 5 has a 52-48 zinc stannate film having a thickness of about $230 \pm 40\text{\AA}$ deposited on the substrate and a zinc oxide film having a thickness of about $80 \pm 40\text{\AA}$ deposited on the 52-48 zinc stannate film. The first titanium primer film on the first silver layer has a thickness of about $18-29\text{\AA}$; the second titanium primer film on the second silver layer has a thickness of about $18-31\text{\AA}$. The remaining layers of Sample 5 are as shown in the Table and have the same composition and thickness for the same films as described in Example 3.

EXAMPLE 6

This Example 6 is Sample 6 of the Table 1 and was made. The coated stack is similar to the coating stack of Example 2 except a zinc oxide film having a thickness of about $80 \pm 40\text{\AA}$ is deposited on the second titanium primer and a 52-48 zinc stannate film having a thickness of $120 \pm 40\text{\AA}$ is deposited on the zinc oxide film. The first titanium primer layer had a thickness of $19-26\text{\AA}$, and the second primer layer had a thickness of $21.5-31\text{\AA}$ and preferably 25\AA . The composition and thickness of the remaining films/layers for Example 6 as shown for Sample 6 in the Table are as described in Example 2.

EXAMPLE 7

This Example 7 is Sample 7 of the Table 1 and was made and is shippable and heatable with the heated coating

stack having reduced haze. The coating stack of Sample 7 is similar to the coating stack of Example 3 except a zinc oxide film having a thickness of about $80 \pm 40\text{\AA}$ was deposited on the second titanium film and a 52-48 zinc stannate film was deposited on the zinc oxide film. The first titanium primer layer had a thickness of about $22\text{-}26\text{\AA}$ and preferably 22.5\AA , and the second titanium primer layer had a thickness of about $18\text{-}25\text{\AA}$ and preferably 21.5\AA . The composition and thickness of the remaining films/layers for Example 7 as shown for Sample 7 on the Table are as described in Example 2.

EXAMPLE 8

This Example 8 is Sample 8 of the Table. The coated article was made and is a coated article suitable for shipment and heat treatment with the heated coated articles having reduced haze. The coated article of this Example 8 is a coated stack deposited on a clear glass substrate. The coating thickness and order of the films is as follows with film 1 deposited on the glass substrate.

TABLE 2

<u>Film No.</u> <u>from Table</u>	<u>Composition of the Film</u>	<u>Thickness of the Film</u>
1	52-48 zinc stannate	$230 \pm 40\text{\AA}$
2	zinc oxide	$80 \pm 40\text{\AA}$
3	1 st silver	$110 \pm 30\text{\AA}$
4	1 st titanium primer	$17\text{-}26\text{\AA}$, preferably 19.5\AA
5	zinc oxide	$80 \pm 40\text{\AA}$
6	52-48 zinc stannate	$740 \pm 40\text{\AA}$
8	2 nd silver film	$110 \pm 30\text{\AA}$
9	2 nd titanium primer	$18\text{-}31\text{\AA}$, preferably 28\AA
10	zinc oxide	$80 \pm 40\text{\AA}$
11	52-48 zinc stannate	$120 \pm 40\text{\AA}$
12	titanium metal overcoat	$29 \pm 3\text{\AA}$

The coated glass having the above coating was used in the fabrication of automotive windshields. The coated glass was cut to size, heated to shape the coated glass, and thereafter laminated to another shaped glass to provide an automotive windshield. The transmittance of the laminate was

DJO 4/29/99 Q.A.M. 4/29/99 DJF 4/29/99

greater than 70%, and ~~low emissivity~~ ^{reflects infrared energy}. The windshield was made as is known in the art. Coated glass for use in automotive windshield was also made substituting 90-10 zinc stannate for the zinc oxide film. The coated article had film thickness in the ranges recited on Table 2.

As can be appreciated, the thickness of the primer layers presented in Table 2 can vary depending on cathode and sputtering equipment. For example, a shippable and heatable coating stack with reduced haze was made with a first titanium primer film having a thickness of $18 \pm 0.5\text{\AA}$ and the second titanium primer film having a thickness of $22 \pm 1\text{\AA}$

EXAMPLE 9

This Example 9 is Sample 9 of the Table and is a coated article that was made. The coated article was shippable and heatable with the heated, coated article having reduced haze. The coated article of this Example 9 is a coated stack deposited on a clear glass substrate. The coating thickness and order of the films is as follows with film 1 deposited on the glass substrate.

TABLE 3

Film No. from Table	Composition of the Film	Thickness of the Film
1	52-48 zinc stannate	$230 \pm 40\text{\AA}$
2	90-10 zinc oxide stannate ^{DJO 4/29/99 DJF 4/29/99}	$80 \pm 40\text{\AA}$
3	1 st silver ^{Q.A.M. 4/29/99}	$107 \pm 30\text{\AA}$
4	1 st titanium primer	$17-24\text{\AA}$, preferably 21.5\AA
5	90-10 zinc stannate	$80 \pm 40\text{\AA}$
6	52-48 zinc stannate	$600 \pm 100\text{\AA}$
7	90-10 zinc stannate	$80 \pm 30\text{\AA}$
8	2 nd silver	$127 \pm 30\text{\AA}$
9	titanium metal primer	$20-26\text{\AA}$, preferably 22.5\AA
10	90-10 zinc stannate	$80 \pm 40\text{\AA}$
11	52-48 zinc stannate	$160 \pm 60\text{\AA}$
12	titanium oxide overcoat	$45 \pm 15\text{\AA}$

As can now be appreciated, a 90-10 zinc stannate film, a zinc oxide film and a zinc oxide, tin oxide film may be interchanged with one another and substituted for one

another to obtain coated articles that are shippable and heatable with reduced haze. However, in the process of the invention the 9-10 zinc stannate film is preferred.

DJO 4/29/99 Q.A.M. 4/29/99 DJF 4/29/99

As can be appreciated, the thickness of the films is not limiting to the invention and may be selected to provide a coated article of a desired color as is known in the art.

Further, the films of all examples of the invention may be

interchanged to attain the features of the invention.

A complete discussion of heating the coated glass sheets to automotive windshields, residential and commercial windows and other transparencies was not given as such technology is known in the art and as can now be appreciated used in the practice

of the invention.

The invention is not limited to the examples presented above and that various changes and alterations can be made without departing from the spirit and broader aspects of the invention, as defined by the claims set forth below and by the range of equivalency allowed by law.

WHAT IS CLAIMED IS:

1. A sputter cathode target comprising:
the target having a 100 weight percent;
5 greater than about 90 and less than 100 weight
percent zinc, and
majority of balance of weight percent tin.
2. The sputter cathode target wherein the weight
10 percent of tin is greater than zero but less than 10 weight
percent.
3. A sputter deposited film comprising:
a dielectric film having greater than about 90 and
15 less than 100 weight percent zinc, and
greater than zero but less than 10 weight percent
tin.
4. An infrared reflective coated article
20 comprising:
a substrate;
a dielectric layer sputter deposited over the
substrate, the layer comprising a first zinc stannate film
deposited over the substrate having zinc in weight percent
25 range of equal to and greater than 10 and equal to and less
than 90, and tin in the weight percent range of equal to and
less than 90 and equal to and greater than 10, and an
electrical enhancing film deposited over the zinc stannate
film, the electrical enhancing film selected from the group of
30 films consisting of zinc oxide, tin oxide film and a second
zinc stannate film wherein the composition of the first zinc
stannate film is at least about 5 weight percent different
than the composition of the second zinc stannate film, and
an infrared reflective layer deposited on the
35 dielectric layer.

5. The coated article of claim 4 wherein the infrared reflective metal is silver and the silver is deposited on the zinc oxide, tin oxide film.

5 6. The article stack of claim 4 wherein the infrared reflective layer is a silver film and the silver film is deposited on the second zinc stannate film.

7. The article stack of claim 4 wherein the dielectric layer is a first dielectric layer and the infrared reflective layer is a first infrared reflective layer and further including:

a metal primer layer over the first infrared reflective layer;

15 a second dielectric layer over the primer layer and optionally a protective overcoat over the second dielectric layer.

8. The article of claim 7 wherein the second dielectric layer is a zinc stannate film having 10-90 weight percent zinc and 90-10 weight percent tin.

9. The article of claim 4 wherein the dielectric layer is a first dielectric layer and the infrared reflective layer is a first infrared reflective layer and further including:

a first metal primer layer over the first infrared reflective metal layer;

30 a second dielectric layer including over the first primer layer;

a second infrared reflective layer over the second dielectric layer;

a second metal primer layer over the second infrared reflective layer;

35 a third dielectric layer including over the second metal primer layer; and

optionally a protective film over the third dielectric layer.

10. The article of claim 9 wherein at least one of
5 the second and third dielectric layers includes a zinc stannate film having 10-90 weight percent zinc and 90-10 weight percent tin.

11. The coating stack of claim 4 wherein the
10 dielectric layer is a first dielectric layer and the infrared reflective layer is a first infrared reflective layer and further including:

a first metal primer layer over the first reflective layer;

15 a second dielectric layer over the first metal primer layer, the second dielectric layer comprising a first dielectric film and a zinc stannate film defined as a first zinc stannate film, the first zinc stannate film having zinc in the weight percent range of equal to and greater than 10
20 and equal to and less than 90 and tin in the weight percent range of equal to and greater than 10 and equal to and less than 90, the first dielectric layer deposited over the first metal primer layer;

a second infrared reflective layer deposited over
25 the second dielectric layer;

a second metal primer layer deposited over the second infrared reflective layer;

a third dielectric layer deposited over the second primer layer; and

30 optionally a protective layer over the third dielectric layer.

12. The coating stack of claim 11 wherein the first dielectric film of the second dielectric layer comprises
35 a zinc oxide film; a zinc oxide, tin oxide film or a zinc stannate film defined as a second zinc stannate film, the second zinc stannate film having a composition different than

the composition of the first zinc stannate film of the second dielectric layer.

13. The coating stack of claim ¹² ~~10~~ wherein the ^{DPO 4/29/99 O.A.M. 4/29/99 JEP 4/29/99}
5 second zinc stannate film of the second dielectric layer has zinc in the weight percent range of equal to and greater than 60 and equal to and less than 90 and tin in the weight percent of equal to and greater than 10 and equal to and less than 40, and the third dielectric layer is a zinc stannate film.

10

14. The coating stack of claim 4 wherein the dielectric layer is a first dielectric layer and the infrared reflective layer is a first infrared reflective layer and further including:

15 a first metal primer layer over the first reflective layer;

a second dielectric layer over the first metal primer film;

20 a second infrared reflective layer over the second dielectric layer;

a second metal primer layer over the second infrared reflecting metal layer;

25 a third dielectric layer over the second metal primer layer, the ~~second~~ ^{third} dielectric layer comprising a first dielectric film and a zinc stannate film defined as a first zinc stannate film, the first zinc stannate film having zinc in a weight percent with the range of equal to and greater than 10 and equal to and less than 90 and tin within the weight percent range of equal to and less than 90 and equal to
30 and greater than 10, the third dielectric film deposited over the second metal primer; and

optionally a protective film overlying the third dielectric film.

35 15. The article of claim 14 wherein the first dielectric film of the third dielectric layer is selected from the group consisting of a zinc oxide film; a zinc oxide, tin

oxide film or a zinc stannate film defined as a second zinc stannate film, the second zinc stannate film having a composition different than the composition of the first zinc stannate film of third dielectric layer.

5

16. The article of claim 15 wherein the second zinc stannate film of the third dielectric layer has zinc in the weight percent range of equal to and greater than 60 and equal to and less than 90 and tin in the weight percent range
10 of equal to and greater than 10 and equal to and less than 40.

17. The article of claim 4 wherein the dielectric layer is a first dielectric layer and the infrared reflective layer is a first infrared reflective layer and further
15 including:

a first metal primer layer over the first reflective layer;

a second dielectric layer over the first metal primer layer, the second dielectric layer comprising a first dielectric film and a ~~first~~ zinc stannate film defined as a first zinc stannate film, the first zinc stannate film having zinc in a weight percent within the range of equal to and greater than 10 and equal to and less than 90 and tin within the weight percent range of equal to and less than 90 and equal to and greater than 10, the second dielectric ~~film~~ layer
20 deposited over the first metal primer layer; ^{210 4/21/99 OAM. 4/29/99 JP 4/28/99}
25 ^{210 4/21/99 OAM. 4/29/99 JP 4/28/99}

a second infrared reflective layer over the first zinc stannate film of the second dielectric layer;

a second metal primer layer over the second infrared
30 reflective layer;

a third dielectric layer over the second metal primer layer, the third dielectric layer comprising a first dielectric film and a zinc stannate film defined as a first zinc stannate film, the first zinc stannate film having zinc
35 in a weight percent within the range of equal to and greater than 10 and equal to and less than 90 and tin within the weight percent range of equal to and less than 90 and equal to

4/29/99

layer D/O 4/29/99
A.M. 4/29/99

and greater than 10, the third dielectric film deposited over the second metal primer layer; and

optionally a protective film overlying the first zinc stannate film of the dielectric layer.

5

18. The coating stack of claim 17 wherein the first dielectric film of the second dielectric layer and the first dielectric film of the third dielectric layer each has a film selected from the group consisting of zinc oxide film; zinc oxide, tin oxide film or second zinc stannate film having a composition different than the composition of the first zinc stannate film in the respective same second or third dielectric layer.

15

19. The coating stack of claim 18 wherein the second zinc stannate film of the first and second dielectric layer each include zinc in the weight percent range of equal to and greater than 60 and equal to and less than 90 and tin in the weight percent of equal to and greater than 10 and equal to and less than 40.

20

20. The coating stack of claim 17 wherein the second dielectric layer further includes a third dielectric film over the first zinc stannate film of the second dielectric layer.

25

21. The coating stack of claim 18 wherein the second dielectric layer further includes a third dielectric film over the first zinc stannate film of the second dielectric layer wherein the third dielectric film of the second dielectric layer is a film selected from the group consisting of zinc oxide film, zinc oxide, tin oxide film and a zinc stannate film defined as a third zinc stannate film, the third zinc stannate film has a composition different than the composition of the zinc stannate film of the second dielectric layer closest to the third zinc stannate film.

30
35

22. The coating of claim 18 wherein the second dielectric film of the ^{second} ~~dielectric~~ layer and the second dielectric film of the third dielectric second layer each comprises a zinc oxide film; a zinc oxide, tin oxide film or a second zinc stannate film having a composition different than the ^{composition} ~~composition~~ of the first zinc stannate film of third dielectric layer.

23. The coating stack of claim ²² ~~220~~ wherein the first and third dielectric films of the second dielectric layer and the first dielectric film of the third dielectric layer each include zinc in the weight percent range of equal to and greater than 60 and equal to and less than 90 and tin in the weight percent of equal to and greater than 10 and equal to and less than 40.

24. The coating stack of claim 20 wherein the second zinc stannate film of the first dielectric layer is on the glass piece and has a thickness in the range of 230 ± 40 Angstroms Å; the first zinc stannate film of the first dielectric layer is on the second zinc stannate film of the first dielectric layer and has a thickness in the range of 80 ± 40 Å; the first infrared reflective metal layer is a first silver film deposited on the first zinc stannate film of the first dielectric layer and has a thickness in the range of 110 ± 30 Å, the metal primer layer is a titanium film deposited on the first silver layer and has a thickness in the range of $17-26$ Å; the first dielectric film of the second dielectric layer is deposited on the titanium film and has a thickness in the range of 80 ± 40 Å; the first zinc stannate film of the second dielectric layer is deposited on the first dielectric film of the second dielectric layer and has a thickness in the range of 740 ± 40 Å; the second infrared reflective metal layer is a second silver film deposited on the second dielectric film of the second dielectric layer and has a thickness in the range of 110 ± 38 Å; the second primer film is a titanium film deposited on the second silver layer and having a thickness in

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the range of 18 - 31Å; the first dielectric film of the third dielectric layer is deposited on the second titanium film and has a thickness in the range of $80 \pm 40\text{\AA}$; the first zinc stannate layer of the third dielectric layer is deposited as on
5 the first dielectric film of the third dielectric layer and has a thickness in the range of $120 \pm 40\text{\AA}$, and the protective layer is a titanium metal film deposited on the first zinc stannate layer of the third dielectric layer and has a thickness in the range of $29 \pm 3\text{\AA}$.

10

25. A coated article comprising:

a substrate;

a first dielectric layer over the substrate;

a first infrared reflective layer over the first

15 dielectric layer;

a first metal primer layer over the first infrared reflective layer;

a second dielectric layer over the first metal primer, the second dielectric layer having a first dielectric film

20 selected from the group consisting of zinc oxide, tin oxide film and a first zinc stannate film, and a second dielectric film the second dielectric film having a composition different than the first dielectric film of the second dielectric layer;

a second infrared reflective layer over the second dielectric layer;

a second primer layer over the second reflective layer;

a third dielectric layer over the second metal primer layer; and

30 optionally a protective layer overlying the third dielectric layer.

26. The coated article of claim 25 wherein the first dielectric layer includes a zinc stannate film, the
35 second dielectric film of the second dielectric layer is a zinc stannate film and the third dielectric layer includes a zinc stannate film, each of the zinc stannate films having

zinc in the weight percent range of 10-90 and tin in the weight percent range of 90-10.

27. The coated article of claim 26 wherein the first dielectric film of the second dielectric layer is the first zinc stannate film having zinc in the weight percent range of equal to and greater than 90 and equal to and less than 60 and tin in the weight percent range of equal to and greater than 10 and equal to and less than 40.

10

28. A coated article comprising:

a substrate;

a first dielectric layer over the substrate;

15 a first infrared reflective layer over the first dielectric layer;

a first metal primer layer over the first infrared reflective layer;

a second dielectric layer over the first metal primer layer;

20 a second infrared reflective layer over the second dielectric layer;

a second metal primer layer over the second reflective metal layer;

25 a third dielectric film having a first dielectric film selected from the group consisting of zinc oxide film; zinc oxide, tin oxide film; a first zinc stannate film and a second dielectric film overlying the first dielectric film, the second dielectric film having a composition different from the first dielectric film; and

30 optionally a protective film overlying the third dielectric layer.

29. The coated article of claim 28 wherein the first and second dielectric layers are each a zinc stannate film, and the second dielectric film of the third dielectric layer is a zinc stannate film and each of the zinc stannate

layer D/10 4/29/99 Q.A.M. 4/29/99 *EP* 4/29/99

films has zinc in the weight percent range of 10-90 and tin in the weight percent range of 90-10.

30. The coated article of claim 29 wherein the
5 first dielectric film of the second dielectric layer has zinc in the weight percent range of equal to and greater than 90 and equal to and less than 60 and tin in the weight percent range of equal to and greater than 10 and equal to and less than 40.

10

31. A coated article comprising:

a substrate;

a first dielectric layer over the substrate;

15 a first infrared reflective layer over the first dielectric layer;

a first primer layer over the first reflective metal layer;

20 a second dielectric layer having a first dielectric film selected from the group consisting of zinc oxide, tin oxide film and a first zinc stannate film, and a second dielectric film overlying the first dielectric film having a composition different than the first dielectric film of the second dielectric layer;

25 a second infrared reflective layer over the second dielectric layer;

a second primer layer over the second reflective layer;

30 a third dielectric layer over the second metal primer layer, the third dielectric layer having a first dielectric film selected from the group consisting of a zinc oxide, tin oxide film and a first zinc stannate film and a second dielectric film, the second dielectric film of the third dielectric layer have a composition different than the composition of the second dielectric film of the third
35 dielectric layer; and

optionally a protective film overlying the third dielectric layer.

32. The coated article of claim 31 wherein the first dielectric layer, the second dielectric film of the second and third dielectric layers are each a zinc stannate film having zinc in the weight percent range of 10-90 and tin in the weight percent range of 90-10.

33. The coated article of claim 32 wherein the first dielectric film of the second and third dielectric layers are each a zinc stannate film having zinc in the weight percent range of equal to and greater than 90 and equal to and less than 60 and tin in the weight percent range of equal to and greater than 10 and equal to and less than 40.

34. The coated article of claim 32 wherein the coated article is a transparency.

35. The coated article of claim 34 wherein the coated article is an automotive transparency.

36. The coated article of claim 35 wherein the automobile transparency is an automotive windshield having a pair of glass sheets laminated together and one of the sheets is fabricated from the substrate having the coating.

37. A method of making an automobile transparency comprising:

applying a coating on a glass substrate having the following:

a first dielectric layer over a glass substrate;

a first infrared reflecting metal layer over the first dielectric film:

a first metal primer layer over the first reflective layer;

a second dielectric layer over the first metal primer layer;

ABSTRACT OF THE DISCLOSURE

Tin is added to a zinc cathode target to enhance the sputter efficiency of the target. Films deposited using the cathode, e.g., greater than zero but less than 10 weight percent tin and greater than 90 but less than 100 weight percent zinc improve the ~~mechanical~~ ^{chemical} durability of a high transmittance, low emissivity coating stack over coating stacks having zinc oxides without tin oxide. High transmittance, low emissivity coating stacks are heated with the heated coating having reduced haze by selecting the thickness of metal primer layer between an infrared reflective film, e.g. a silver film and a dielectric film, e.g., a 52-48 zinc stannate, zinc oxide, tin oxide film or a zinc oxide film. Also disclosed are enhancing films that lower the resistivity of silver films deposited thereon and improve chemical durability of the coating stack.

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DECLARATION FOR PATENT APPLICATION

Docket Number (Optional)
1375A1

As a below named inventor, I hereby declare that:

My residence, post office address and citizenship are as stated below next to my name.

I believe I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural names are listed below) of the subject matter which is claimed and for which a patent is sought on the invention entitled **SHIPPABLE HEAT-TREATABLE SPUTTER COATED ARTICLE AND ZINC CATHODE SPUTTERING TARGET CONTAINING LOW AMOUNTS OF TIN**, the specification of which is attached hereto unless the following box is checked:

☐ was filed on _____ as United States Application Number or PCT International Application Number _____ and was amended on _____ (if applicable).

I hereby state that I have reviewed and understand the contents of the above-identified specification, including the claims, as amended by any amendment referred to above

I acknowledge the duty to disclose information which is material to patentability as defined in Title 37, Code of Federal Regulations, § 1.56.

I hereby claim foreign priority benefits under Title 35, United States Code, § 119(a)-(d) of any foreign application(s) for patent or inventor's certificate listed below and have also identified below any foreign application for patent or inventor's certificate having a filing date before that of the application on which priority is claimed.

Prior Foreign Application(s)

Priority Claimed

☐ Yes ☐ No

(Number)	(Country)	(Day/Month/Year Filed)
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☐ Yes ☐ No

(Number)	(Country)	(Day/Month/Year Filed)
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I hereby claim the benefit under Title 35, United States Code, § 119(c) of any United States provisional application(s) listed below.

60/084,720 May 8, 1998

(Application Number) (Filing Date)

60/085,129 May 12, 1998

(Application Number)	(Filing Date)
1	1/1/2018
2	1/1/2018
3	1/1/2018
4	1/1/2018
5	1/1/2018
6	1/1/2018
7	1/1/2018
8	1/1/2018
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10	1/1/2018
11	1/1/2018
12	1/1/2018
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94	1/1/2018
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96	1/1/2018
97	1/1/2018
98	1/1/2018
99	1/1/2018
100	1/1/2018

I hereby claim the benefit under Title 35, United States Code, § 120 of any United States application(s) listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States application in the manner provided by the first paragraph of Title 35, United States Code, § 112, I acknowledge the duty to disclose information which is material to patentability as defined in Title 37, Code of Federal Regulations, § 1.56 which became available between the filing date of the prior application and the national or PCT International filing date of this application

[illegible]

(Application Number)	(Filing Date)	(Status - patented, pending, abandoned)
100-100000	10/1/1990	patented
100-100001	10/1/1990	pending
100-100002	10/1/1990	abandoned

I hereby appoint the following attorney(s) and/or agent(s) to prosecute this application and to transact all business in the Patent and Trademark Office connected therewith.

Donald C. Lepiane, Registration No. 25,996

Kenneth J Stachel, Registration No. 27,722

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Address all correspondence to: PPG INDUSTRIES, INC.
Intellectual Property Department
One PPG Place
Pittsburgh, Pennsylvania 15272

I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

Full name of sole or first inventor (given name, family name): Dennis J. O'Shaughnessy

Inventor's signature D. J. O'Shea Date 4/29/99

Residence	Allison Park, Pennsylvania, United States of America	Citizenship	United States of America
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Full name of second joint inventor, if any (given name, family name): James J Finley

Inventor's signature James H. Wiley Date 4/29/99

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SHEET 2

Docket Number (Optional)
1375A1

Full name of third joint inventor, if any (given name, family name): Paul A. Medwick
 Second Inventor's signature [Signature] Date 4/27/99
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